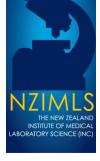
# EXAMINATION FOR QUALIFIED MEDICAL LABORATORY TECHNICIAN



**Candidate Name:** 

**Candidate Number:** 

Subject: BIOCHEMISTRY

**Examination Date:** 8 October 2022

Time Allowed: 3 hours – 9.30am – 12.40pm

10 minutes extra time for reading the paper

#### **General Instructions**

1. Total marks for paper = 100.

2. Marks for each question are as indicated,

3.	The paper consists of:	Common	Discipline Specific
	Section A, questions 1-30 = Total Marks 15	6 Marks	9 Marks
	Section B, questions 31-38 = Total Marks 10	5 Marks	5 Marks
	Section C, questions 39-41 = Total Marks 10	4 Marks	6 Marks
	Section D, questions 42-45 = Total Marks 05	5 Marks	0
	Section E, questions 46-61 = Total Marks 40	10 Marks	30 Marks
	Section F. guestions 62-63= Total Marks 20	0	20 Marks

- 4. All questions are to be attempted.
- 5. Use of calculator is permitted.
- 6. Write all answers into this booklet.

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WORD DEFINITIONS				
Calculate	Perform a mathematical process to get the answer			
Classify	Be able to designate to a group			
Compare	Detail both the differences and the similarities			
Complete	Finish, have all the necessary parts			
Convert	Express in alternative units			
Define	State meaning clearly and concisely			
Describe	Give a complete account demonstrating a thorough practical knowledge			
Discuss	Give details, explaining both the positives and negatives			
Distinguish	To briefly point out the main differences			
Expand	To express at length or in a greater detail			
Identify	Recognise according to established criteria			
Indicate	Briefly point out			
Interpret	Express the results of a test or series of tests in a meaningful format			
Label	Give a name to			
List	Headings only			
Match	Find one that closely resembles another			
Name	A word or group of words used to describe or evaluate			
Outline	Write brief notes incorporating the essential facts			
State	Give the relevant points briefly			

#### **SECTION A**

#### Section A - Question 1 to Question 30 = Total Marks: 15

Multi choice questions

#### Multi choice questions – choose one answer for each question

(0.5 mark per correct answer)

#### Circle the letter for the correct answer

**Example**. Which of the below is a primary colour?

- a. Green
- b. Purple
- C.

Red

Orange

- C.1 Approximately what percent alcohol is in a standard use hand sanitiser?
  - a. 95%
  - b. 75%
  - c. 30%
  - d. 10%
- C.2 The patella is part of which human joint?
  - a. Shoulder
  - b. Elbow
  - c. Knee
  - d. Wrist
- C.3 An anticoagulant is used to:
  - a. stop blood clotting
  - b. stop blood haemolysing
  - c. help blood separating
  - d. separate red cells and plasma

C.4	ch of the following is <b>NOT</b> listed in the Health and Safety at Work Act 2015 as "Duties of kers"?				
	a.	take reasonable care for his or her own health and safety			
	b.	take reasonable care that his or her acts or omissions do not adversely affect the health and safety of other persons			
	c.	co-operate with any reasonable policy or procedure of the PCBU (person conducting a business or undertaking) relating to Health and Safety at the workplace that has been notified to workers			
	d.	issue provisional improvement notices			
C.5	Which of the following statements is true of an acidic solution?				
	a.	has a pH less than 7			
	b.	is caustic			
	C.	has a pH greater than 7			
	d.	is Isotonic			
C.6	The reference interval for a given test is based on the results that are seen in what percent of the healthy population?				
	a.	5%			
	b.	10%			
	c.	90%			
	d.	95%			
C.7	Trea	ating all blood and body fluids as potentially infectious is an example of:			
	a.	Laboratory standard operating procedures			
	b.	CDC guidelines			
	C.	Standard precautions			
	d.	Health and safety requirements			
C.8	Which laboratory department is primarily responsible for the diagnosis of leukaemia?				
	a.	Haematology			
	b.	Histology			
	C.	Blood Transfusion			

d.

Biochemistry

C.9	Hormones are produced by which bodily system?				
	a.	Lymphatic			
	b.	Cardiovascular			
	c.	Endocrine			
	d.	Digestive			
C.10	Form	alin is a laboratory fluid used to			
	a.	Preserve tissue samples			
	b.	Wash histology cutting knives			
	c.	Clean benches			
	d.	Decontaminate centrifuges			
C.11	A chemical that is described as a carcinogen poses what specific risk?				
	a.	It may burn the skin			
	b.	It may cause cancer			
	c.	It may poison the liver			
	d.	It may cause loss of vision.			
C.12	The p	ractice of enforcing document management standards within the workplace is referred			
	a.	Quality management			
	b.	Quality control			
	c.	IANZ requirements			
	d.	Document control			
D.13		h of the following analytes can show falsely increased results if the sample has been minated with heparin anticoagulant?			
	a.	Potassium			
	b.	Lithium			
	c.	Calcium			
	d.	Glucose			

C.9

D.14 When a SST sample is contaminated with EDTA anticoagulant, which of the follow can falsely be increased?				
	a.	Sodium		
	b.	Calcium		
	c.	Glucose		
	d.	Potassium		
D.15	Hypol	kalemia is a decrease in which blood parameter?		
	a.	Potassium		
	b.	Sodium		
	c.	Calcium		
	d.	Magnesium		
D.16	What	major blood vessel carries blood back to the heart from the body?		
	a.	The vena cava		
	b.	The pulmonary artery		
	c.	The aorta		
	d.	The iliac artery		
D.17	Systol	e refers to which of the following terms?		
	a.	Refilling of the ventricle immediately after ventricular contraction		
	b.	Contraction of the ventricle of the heart		
	c.	Contraction of the atrium of the heart		
	d.	Contraction of aorta		
D.18	Elevat	ted serum potassium levels can be associated with which of the following conditions?		
	a.	Hepatitis		
	b.	Dehydration		
	c.	Myeloma		
	d.	Congestive heart failure		
D.19	Overd	lose of paracetamol can result in which condition?		
	a.	Ketoacidosis		
	b.	Osteoporosis		
	C.	Liver failure		
	d.	Gout		

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D.20	In a myocardial infarction, which blood parameter is likely to be increased?			
	a.	Bicarbonate		
	b.	Iron		
	c.	Free triiodothyronine		
	d.	Troponin		
D.21		n using a mechanical pipette, which of the following refers to a forward pipetting sique?		
	a.	first stop, aspirate, dispense all contents		
	b.	second stop, aspirate, dispense all contents		
	c.	first stop, aspirate, dispense to first stop		
	d.	second stop, aspirate, dispense to first stop		
D.22	Wher	e are the adrenal glands located?		
	a.	In the middle of the neck		
	b.	On top of the kidneys		
	c.	Below the brain		
	d.	Adjacent to the pancreas		
D.23	What	is the function of the villi?		
	a.	Filter toxins from the blood		
	b.	Hormone regulation		
	c.	Facilitate gas exchange		
	d.	Absorption of nutrients		
D.24	If a pa	aediatric patient has severe lipaemia, what colour can the serum appear?		
	a.	Red		
	b.	Yellow		
	c.	Green		
	d	White		
D.25		ST sample has been exposed to direct sunlight for a long period, which of the following neter is falsely decreased?		
	a.	Potassium		
	b.	Magnesium		
	c.	Bilirubin		
	d.	Chloride		

D.26	In hereditary haemochromatosis, which of the following analyte is likely to be raised?				
	a.	Insulin			
	b.	Copper			
	C.	ACTH			
	d.	Iron			
D.27	Eleva	ted creatinine in blood can be attributed to which of the following condition?			
	a.	Heart conditions			
	b.	Liver cirrhosis			
	c.	Renal failure			
	d.	Liver cholestasis			
D.28	Cocai	ne is routinely analysed by which principle?			
	a.	Colourimetry			
	b.	HPLC			
	c.	Immunochemistry			
	d.	Electrophoresis			
D.29	An elevated gamma fraction in a monoclonal band for protein electrophoresis suggests which of the following?				
	a.	Renal failure			
	b.	Haemolysis			
	c.	Autoimmune disease			
	d.	Brain tumour			
D.30	An ov	verdose of Digoxin can result in which condition?			
	a.	Irregular heartbeats			
	b.	Renal failure			
	С.	Cerebral haemorrhage			
	d.	Pulmonary embolism			
	۵.				

**END OF SECTION** 

#### **SECTION B**

Labelling of diagrams e.g. anatomy, hazard identification, instrument

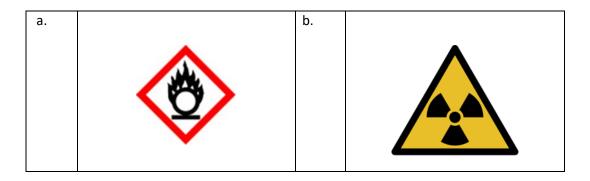
#### Section B - Question 31 to Question 38 = Total Marks: 10

(Answer all questions)

#### C.31 Name the following hazard symbols

(0.5 marks per correct answer)

(C.31: 1 mark)

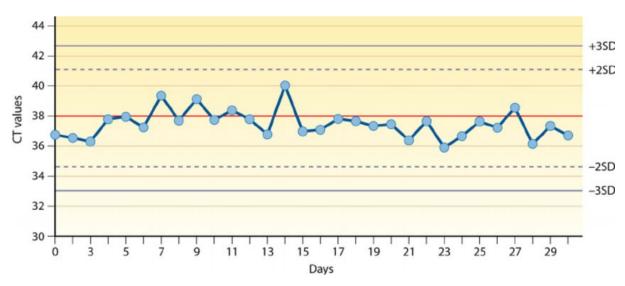


- a. \_\_\_\_\_
- b. \_\_\_\_\_

#### C.32 Name the type of graph:

(0.5 marks per correct answer)

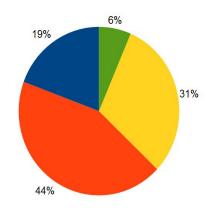
(C.32: 1.5 marks)



Type of graph:

Name the axis: CT values = axis

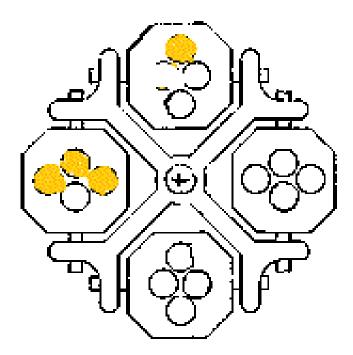
Days = \_\_\_\_\_ axis



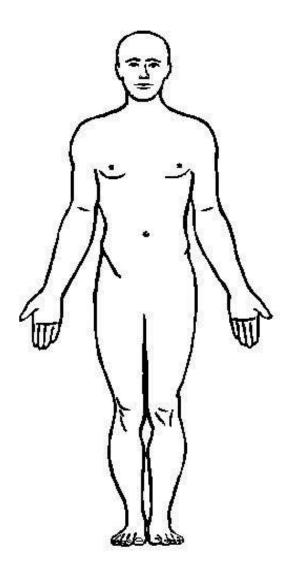
C.34 The yellow dots represent blood tubes in a swing out centrifuge rotor, assume all tubes are filled to the same level.

You have 4 more tubes to centrifuge, indicate on the rotor where they need to be positioned.

(C.34: 0.5 mark)



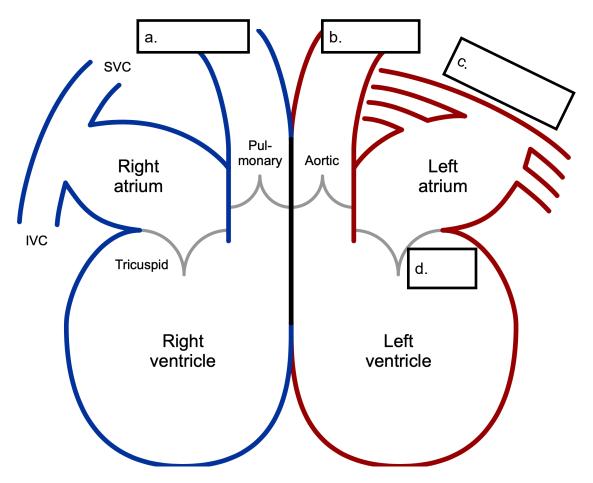
- a. Femoral artery
- b. Jugular vein
- c. Median cubital vein



D.36 On the diagram, add the correct labels to the boxes a.-d.

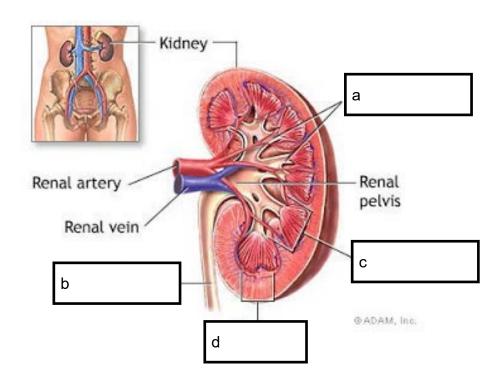
(0.5 marks per correct answer)

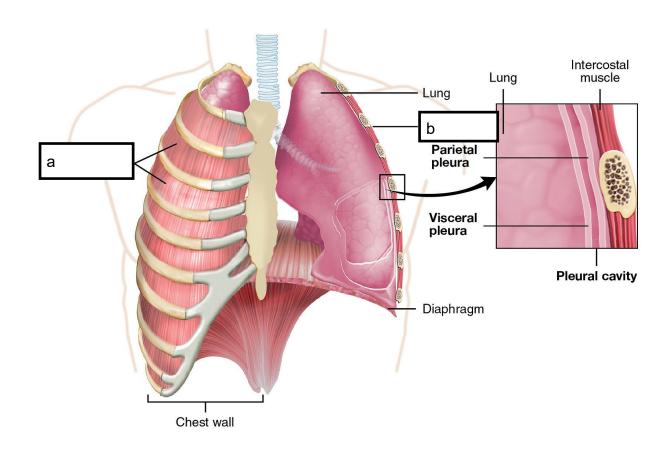
(D.36: 2 marks)



D.37 On the diagram, add the correct labels to the boxes a.-d.

(0.5 marks per correct answer)
(D.37: 2 marks)





#### **END OF SECTION**

#### **SECTION C**

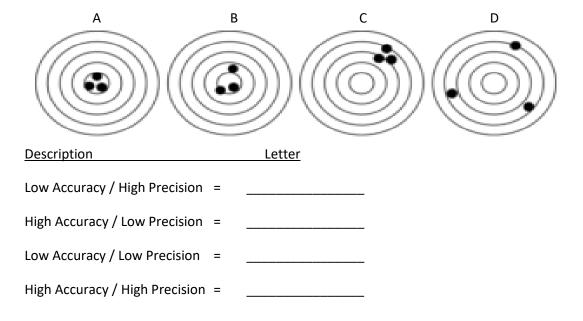
#### Tables, match column definition

#### Section C - Question 39 to Question 41 = Total Marks: 10

#### (Answer all questions)

C.39 Select the correct letter for each description:

(C.39: 2 marks)



C.40 Match Column A to Column B, and write your answers in the table below: (Roman Numerals only required): (C.40: 2 marks)

Column A	Column B
Medical Sciences Council	i. Certifies laboratory quality systems
International Accreditation New Zealand	ii. Issues Annual Practising Certificate
New Zealand Institute of Medical Laboratory Science	iii. Patients' rights for service
Health and disability commission	iv. Professional affairs and education

Column A	Column B (Roman numerals only required)
Medical Sciences Council	
International Accreditation New Zealand	
New Zealand Institute of Medical Laboratory Science	
Health and disability commission	

D.41 Match the laboratory test (Alpha) to the correct description (Numeric), and write your answers in the table below (Roman Numeral only required): (0.5 marks per correct answer)

(D.41: 6 marks)

	Laboratory Test		Description
а	Calcium	i	Increased in metabolic alkalosis
b	Urea	ii	Used for cardiovascular risk assessment
С	Gamma glutamyl-transferase	iii	Increased in gout
d	Lipase	iv	Falsely increased in haemolysis
е	Lactate Dehydrogenase	v	Increased in dehydration
f	Beta-Hydroxybutyrate	vi	Increased in tumours
g	Bicarbonate	vii	Increased in acute pancreatitis
h	Triglycerides	viii	Decreased in anaemia
i	Iron	ix	Increased in post hepatic obstruction
j	Uric Acid	x	Decreased in hypoparathyroidism
k	Beta hCG	xi	Increased in diabetic ketoacidosis
I	Salicylate	xii	Can cause anaphylaxis reactions

	Laboratory Test	Description
а	Calcium	
b	Urea	
С	Gamma glutamyl-transferase	
d	Lipase	
е	Lactate Dehydrogenase	
f	Beta-Hydroxybutyrate	
g	Bicarbonate	
h	Triglycerides	
i	Iron	
j	Uric Acid	
k	Beta hCG	
I	Salicylate	

#### **END OF SECTION**

#### **SECTION D**

#### Calculations

# Section D – Question 42 to Question 45 = Total Marks: 5

## **Calculations**

	A Glucose Tolerance Test dose is 75g glucose in 350mL water. This test requires the fast for 12 hours before drinking the solution. A blood test is then collected 120 min the drinking the solution. (C.42: 1.5)							
a. Calculate the percentage glucose in solution. (Show working)								
a								
b. If the patient finished their evening meal at 2115 hrs, state the earliest time they can present for the test the following day. (0.5								
)								
				e the blood test	is required. (0.5 mark)			
Refer to daily	fridge temper	ature monitorin	g record below.		(C.43: 1 mark)			
ay of the eek	Monday	Tuesday	Wednesday	Thursday	Friday			
aily Fridge emperature.	4.6	3.8	3.1	9.3	5.1			
a. Calculate t	he mean reco	rded temperatu	re for the week. (	Show calculatio	ns)			
i								
	Refer to daily ay of the eek and Fridge emperature.	ay of the mean reco	D. If the patient finished their evening mea present for the test the following day.  D	Defer to daily fridge temperature monitoring record below.  All of the Monday Tuesday Wednesday eek ally Fridge 4.6 3.8 3.1 3.1 3.1 3.1 3.1 3.1 3.1 3.1 3.1 3.1	D. If the patient finished their evening meal at 2115 hrs, state the earliest time present for the test the following day.  D			

C.44	Convert the following:			(C.44: 1.5 marks)
	a) 4.5 mL	to	μL	
	b) 1.125kg	to	g	
	c) 1500 µmol	to	mmol	
C.45	Calculate how many gra solution?	ms of so	odium chloride (NaCl) a	re required to make 1.0L of a 2 Molar
	(Show calculations) Atomic Weight of sodiun Atomic Weight of chlorin			(C.45: 1 mark)

**END OF SECTION** 

#### **SECTION E**

Short answer questions (answers = one or more words, short sentences)

# Section E – Question 46 to Question 61 = Total Marks: 40

## **Short Answer Questions**

C.46	List the activities that registered laboratory staff must do to comply with the HPCA  (C					
C.47	Outline cultural competency as it relates to medical laboratory science?	(C.47: 2 marks				
C.48	Outline Total Quality Management in the medical laboratory setting	(C.48: 2 marks				

C.49	Describe the procedures taken when dealing with a blood spill in the laboratory or
	phlebotomy clinic? (C.49: 2 marks)
C.50	Define Occupational Overuse Syndrome in a medical laboratory workplace. Name a common cause and who should you speak to if you suffer from it? (C.50: 1.5 marks)
C.51	On removing a reagent or product from a laboratory fridge, it is found to be a room
	temperature. What is the correct process to follow? (C.51: 1.5 marks)
	<del></del>

D.52	Disti	nguish between plasma and serum samples:	(D.52: 2 marks)
D.53	Desc syste	eribe the basic function of the following systems, and name 2 or em: (1 mark per correct function, and 0.5 mark)	rgans involved in each
	5,500	(=	(D.53: 6 marks)
	a.	Gastrointestinal	
	b	Cardiovascular	
	C.	Endocrine	

D.54	4 Distinguish between the principles of operation in colourimetry, immunoche	
	ion-specific electrode systems:	(D.54: 3 marks)
		<i>(</i> 1
D.55	5 Distinguish between spectrophotometric end point assays and rate reaction	(kinetic) assays
		(D.55: 2 marks
D.56	Distinguish between venous, arterial, and capillary samples. Your answer sh oxygen level in each sample type:	ould include the <b>(D.56:3 marks</b> )

a. <i>i</i>	Accuracy and Precision
b.	Forward and Reverse Pipetting
c.	Direct and Indirect ISE Measurement
	oe two uses and the mode of action for potassium oxalate with sodium fluoride (fluoride) grey top tubes: (0.5 marks per correct use; and 0.5 marks for each mode of action
	(D.58: 2 marks
	c. Descril

neasurement of pH:	H measurement, and describe	e at least one method for the <b>(D.59:</b>
		(D.60:
Chemistry	Result	Reference range
<b>Chemistry</b> Sodium	Result 136 mmol/L	Reference range
-		
Sodium	136 mmol/L	135-145 mmol/L
Sodium Potassium	136 mmol/L 3.9 mmol/L	135-145 mmol/L 3.6-5.2 mmol/L
Sodium Potassium Calcium	136 mmol/L 3.9 mmol/L 2.3 mmol/L	135-145 mmol/L 3.6-5.2 mmol/L 2.1-2.55 mmol/L

D.61 Interpret the following blood results from an adult female, state which organ is in disorder and the treatment for it: (1 mark for the correct organ and 1 mark for the treatment)

(D.61: 2 marks)

Chemistry	Result	Reference range
Sodium	156 mmol/L	135-145 mmol/L
Potassium	6.1 mmol/L	3.6-5.2 mmol/L
Calcium	2.0 mmol/L	2.1-2.55 mmol/L
Albumin	25 mmol/L	32-48 mmol/L
Creatinine	298 umol/L	45-90 umol/L
Troponin	11 ng/L	<15 ng/L

#### **END OF SECTION**

#### **ESSAY**

# Section F – Question 62 to Question 63 = Total Marks: 20

# **Essay Questions**

(D.62: 10 r	case.


-	In essay format, describe pre-analytical e collection, in relation to biochemistry, an	d how best to prevent	them. ( <b>D.63: 10 mar</b>


#### **EXTRA PAPER**


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